

PHYS 2170
General Physics 3 for Majors
Fall 2021

Lecture 36

Angular momentum in three dimensions

November 19

1 Review the central force problem in two dimensions.



$$E = \frac{\vec{p}^2}{2M} + V(r)$$

In classical mech: $E = \frac{1}{2}mv_r^2 + \frac{L_z^2}{2Mr^2} + V(r)$

(effectively 1d)

effective potential

In QM: look for states of energy E :

$$E\Psi = -\frac{\hbar^2}{2M} \left[\frac{1}{r} \frac{\partial}{\partial r} [r\Psi] \right] + \left(V(r) + \frac{(mr)^2}{2Mr^2} \right) \Psi$$

Angular momentum

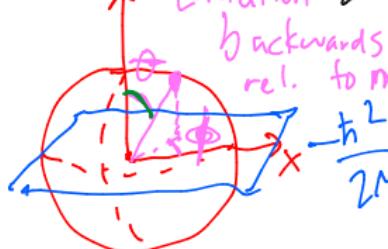
$$L_z = m\hbar$$

$$m = 0, \pm 1, \pm 2, \dots$$

2

State the solution to the central force problem in three dimensions in quantum mechanics.

$$\text{In 3d : } -\frac{\hbar^2}{2M} \left[\frac{\partial^2 \Psi}{\partial x^2} + \frac{\partial^2 \Psi}{\partial y^2} + \frac{\partial^2 \Psi}{\partial z^2} \right] + U(r) \Psi = E \Psi$$

 [notation] backwards
rel. to math] [spherical coords. (ugly!)]

$$-\frac{\hbar^2}{2M} \left[\frac{1}{r^2} \frac{\partial}{\partial r} (r^2 \Psi) \right] + \left(U(r) + \frac{\hbar^2 L^2}{2Mr^2} \right) \Psi = E \Psi$$

$$\begin{aligned} \vec{\nabla}^2 \Psi &= \hbar^2 \left[-\frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial \Psi}{\partial \theta} \right) \right. \\ &\quad \left. - \frac{1}{\sin^2 \theta} \frac{\partial^2 \Psi}{\partial \phi^2} \right] \end{aligned}$$

$$\vec{\nabla}^2 = L_x^2 + L_y^2 + L_z^2$$

Try separation of variables: $\Psi = R(r) Y_1(\theta) Y_2(\phi)$

3 Describe the quantization of \mathbf{L}^2 and L_z in quantum mechanics.

$$\Psi = R(r) \underbrace{Y_1(\theta)}_{\bullet} Y_2(\phi) \dots$$

$$\bullet Y_l(\theta) = f_{lm}(\theta)$$

$$\bullet Y_2(\phi) = e^{im\phi} \quad (m=0, \pm 1, \pm 2, \dots)$$

$$l=0, 1, 2, \dots$$

$$\hookrightarrow L_z = m\hbar$$

$$(\vec{L}^2 \geq l_z^2) \rightarrow \underline{l \geq |m|}$$

How many allowed states have $l=3$?

- $3 \geq |m|$
 - $m = -3, -2, -1, 0, 1, 2, 3$
 - 7 possible choices
- In general, when $l > 0$,
degeneracy of $2l+1$

$$\bullet \vec{L}^2 \Psi = \hbar^2 l(l+1) \Psi$$

• total angular momentum
is quantized

The effective radial potential:

$$V_{\text{eff}}(r) = V(r) + \frac{\hbar^2 l(l+1)}{2Mr^2}$$

depends only on l

4

A rotating molecule (in ~~three~~ dimensions) has energy levels

$$E \Psi = \frac{\hbar^2}{2I} \Psi \quad \left[\text{rotational} \quad \frac{\vec{p}^2}{2M} \right]$$

What are the quantized energy levels? What is the degeneracy of the 3rd lowest energy level?

Allowed energy levels : $E_l = \frac{\hbar^2 l(l+1)}{2I}$ $l=0,1,2,\dots$

3rd lowest energy : $l=2$: $E_2 = \frac{\hbar^2 \cdot 2(2+1)}{2I} = 3 \frac{\hbar^2}{I}$

degeneracy : 5

$$L_z = -2\hbar, -\hbar, 0, \hbar, 2\hbar$$

5

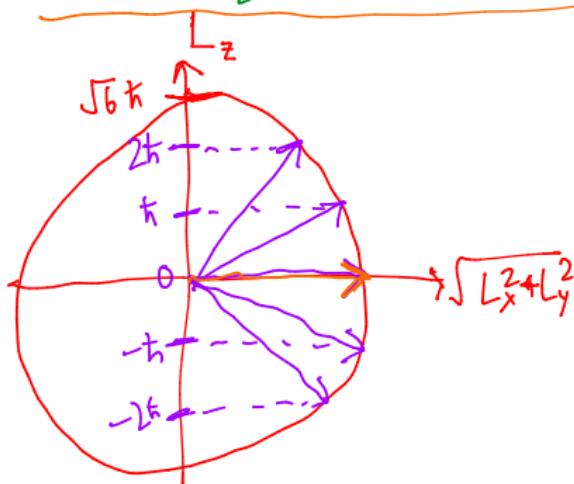
How is it possible to have $\mathbf{L}^2 > \max(L_z^2)$?

Take $l=2$ states:

$$\vec{l}^2 = \hbar^2 l(l+1) = \hbar^2 \cdot 2 \cdot (2+1) = 6\hbar^2$$

Max value of L_z^2 : ($m=2$): $(2\hbar)^2 = L_z^2 = 4\hbar^2$

$$\vec{l}^2 - L_z^2 = 6\hbar^2 - 4\hbar^2 = 2\hbar^2 = \hbar^2 + \hbar^2.$$



max value of L_x, L_y, L_z
should all be $2\hbar$
($l\hbar$)

Heisenberg!:

$$\Delta L_x \Delta L_y \geq \frac{\hbar}{2} \langle L_z \rangle$$

Look at $m=2$:

$$\Delta L_x \Delta L_y \geq \frac{\hbar}{2} \cdot 2\hbar = \hbar^2$$

Guess: $\Delta L_x = \Delta L_y$ $\Delta L_x^2 = \Delta L_y^2 = \hbar^2$.