

PHYS 4410
Quantum Mechanics 2
Spring 2023

Lecture 23

The Born-Oppenheimer approximation

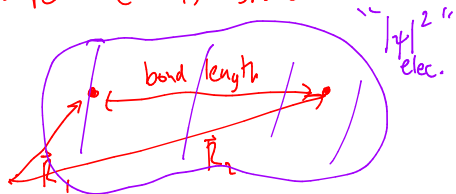
March 15

1 Today: covalent bond / hybrid orbitals

Eg. H_2^- (hydrogen molecule ion):

$$H = \underbrace{\frac{\vec{p}_1^2}{2M} + \frac{\vec{p}_2^2}{2M}}_{\text{proton: mass } M} + \frac{e^2}{4\pi\epsilon_0 |\vec{R}_1 - \vec{R}_2|} + \underbrace{\frac{\vec{p}^2}{2m}}_{\substack{\text{mass } m \\ \text{of electron}}} - \frac{e^2}{4\pi\epsilon_0 |\vec{r} - \vec{R}_1|} - \frac{e^2}{4\pi\epsilon_0 |\vec{r} - \vec{R}_2|}$$

Expect: e^- is "shared" btwn two protons



Key idea:

large dimensionless...

$$\frac{M}{m} \sim 1800$$

—————

2 If $M \rightarrow \infty$... then

$$H_{el} = \frac{e^2}{4\pi\epsilon_0 |\vec{R}_1 - \vec{R}_2|} + \frac{\vec{p}^2}{2m} - \frac{e^2}{4\pi\epsilon_0 |\vec{r} - \vec{R}_1|} - \frac{e^2}{4\pi\epsilon_0 |\vec{r} - \vec{R}_2|}$$

\vec{R}_1, \vec{R}_2 : fixed constants.

simultaneously diagonalize

H_{el} does not depend on \vec{P} :

$$[H_{el}, \vec{R}_1] = [H_{el}, \vec{R}_2] = 0$$

$$[\vec{R}_1, \vec{R}_2] = 0$$

Born-Oppenheimer approximation:

Step 1:

$$E_{eff}(\vec{R}_1, \vec{R}_2) \psi_{el}(\vec{r}; \vec{R}_1, \vec{R}_2) = H_{el}(\vec{R}_1, \vec{R}_2) \psi_{el}(\vec{r}; \vec{R}_1, \vec{R}_2) \quad (\vec{R}_1, \vec{R}_2)$$

Step 2: $H = \frac{\vec{P}_1^2}{2M} + \frac{\vec{P}_2^2}{2M} + H_{el} \dots$

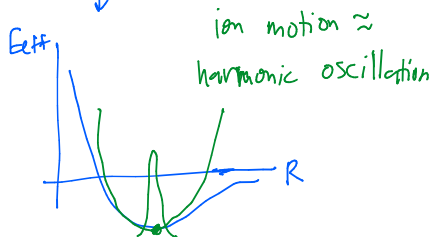
$$\psi_{tot}(\vec{r}, \vec{R}) = \psi_{ion}(\vec{R}) \psi_{el}(\vec{r}; \vec{R})$$

$$H[\psi_{ion} \psi_{el}] = [E_{eff}(\vec{R}) \psi_{el}] \psi_{ion} + \left[\frac{\vec{P}_1^2}{2M} + \frac{\vec{P}_2^2}{2M} \psi_{ion} \right] \psi_{el}$$

$$\cancel{\psi_{ion} \left[\frac{\vec{P}_1^2}{2M} + \frac{\vec{P}_2^2}{2M} \right] \psi_{el}} \rightarrow \left[\frac{\partial}{\partial \vec{R}} \psi_{el} \right] \frac{1}{M} \sim \frac{1}{a_B} \frac{1}{M} \sim \frac{m}{M} \ll 1$$

$$\boxed{3} \quad H \psi_{el} \psi_{ion} = \underbrace{\left[\frac{\vec{p}_1^2 + \vec{p}_2^2}{2m} \psi_{ion} \right]}_{\text{Equations for just ions!}} \cdot \psi_{el} + \left(E_{eff}(\vec{R}) \psi_{ion} \right) \cdot \psi_{el} = E \psi_{ion} \psi_{el}$$

Because ions have large $M \dots$



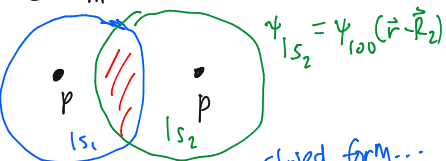
Interesting dynamics / ψ is associated to ψ_{el}

Main problem: $H_{el} \psi_{el} = \underbrace{E_{eff}(\vec{R})}_{\text{classical problem: find minima of } E_{eff}(\vec{R})} \psi_{el}$

classical problem: find minima of $E_{eff}(\vec{R})$

4 Even He_2 can't be solved exactly...

Common ansatz: Linear Combo of Atomic Orbitals (LCAO)



$$|\sigma_{\pm}\rangle = A \left[|1s_1\rangle \pm |1s_2\rangle \right]$$

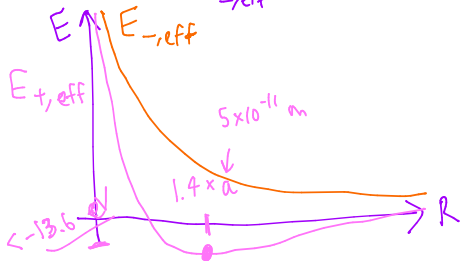
$\psi_{1s_i} = \psi_{100}(\vec{r} - \vec{r}_i)$ ← closed form...

$n=1, l=0, m=0$

Caveat: $A \neq 1/\sqrt{2}$ b/c $\langle 1s_1 | 1s_2 \rangle \neq 0$.

Variational methods: $E_{\pm, \text{eff}}(R) = \langle \sigma_{\pm} | \text{He}_2 | \sigma_{\pm} \rangle$

[analytically done
in Tong...
follow lec 2]



negative energy... stable
covalent bond

5 What happens if $\text{H}_2^- \rightarrow \text{H}_2$ (2 protons / 2 e1)?

Proceed w/ variational calculation:

$$|\psi\rangle = |\sigma_+ \sigma_+\rangle \otimes \frac{|\uparrow\downarrow\rangle - |\downarrow\uparrow\rangle}{\sqrt{2}}$$

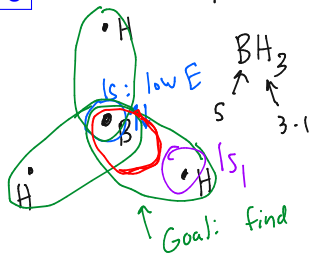
pick best value of R

true: $\approx -0.17 \text{ eV}$

Calculate $\langle \psi | H_{\text{el}} | \psi \rangle \approx \underbrace{-0.1 \text{ eV}} - 2 \cdot 13.6 \text{ eV}$

more stable to have H_2 rather than $\text{H} + \text{H}$

6 Much of quantum chemistry ... follows from these ideas.



$$b = 3 \cdot 2$$

8 unpaired electrons...

$$\frac{|p_x\rangle \pm i|p_y\rangle}{\sqrt{2}} = \begin{pmatrix} n \\ 2 \\ \pm 1 \\ m \end{pmatrix}$$

Goal: find LCAO ansatz for 3 covalent bonds.

Theory of symmetry: "sp² hybrid orbital" $\begin{cases} |2sp_1^2\rangle = a_1|2s\rangle + a_2|2p_x\rangle \\ |2sp_{2,3}^2\rangle = a_1|2s\rangle + a_2\left[\pm\frac{\sqrt{3}}{2}|2p_y\rangle - \frac{1}{2}|2p_x\rangle\right] \end{cases}$

Like before: $|\sigma_{1,2,3}\rangle = A_1|2sp_{1,2,3}^2\rangle + A_2|1s_{1,2,3}\rangle$

Lowest energy trial: $|\psi\rangle \sim$ antisymmetric... $|1s_B\rangle \otimes |1s_B\rangle \otimes |\sigma_1\rangle \otimes |\sigma_1\rangle \otimes |\sigma_2\rangle \otimes |\sigma_2\rangle \otimes |\sigma_3\rangle \otimes |\sigma_3\rangle$